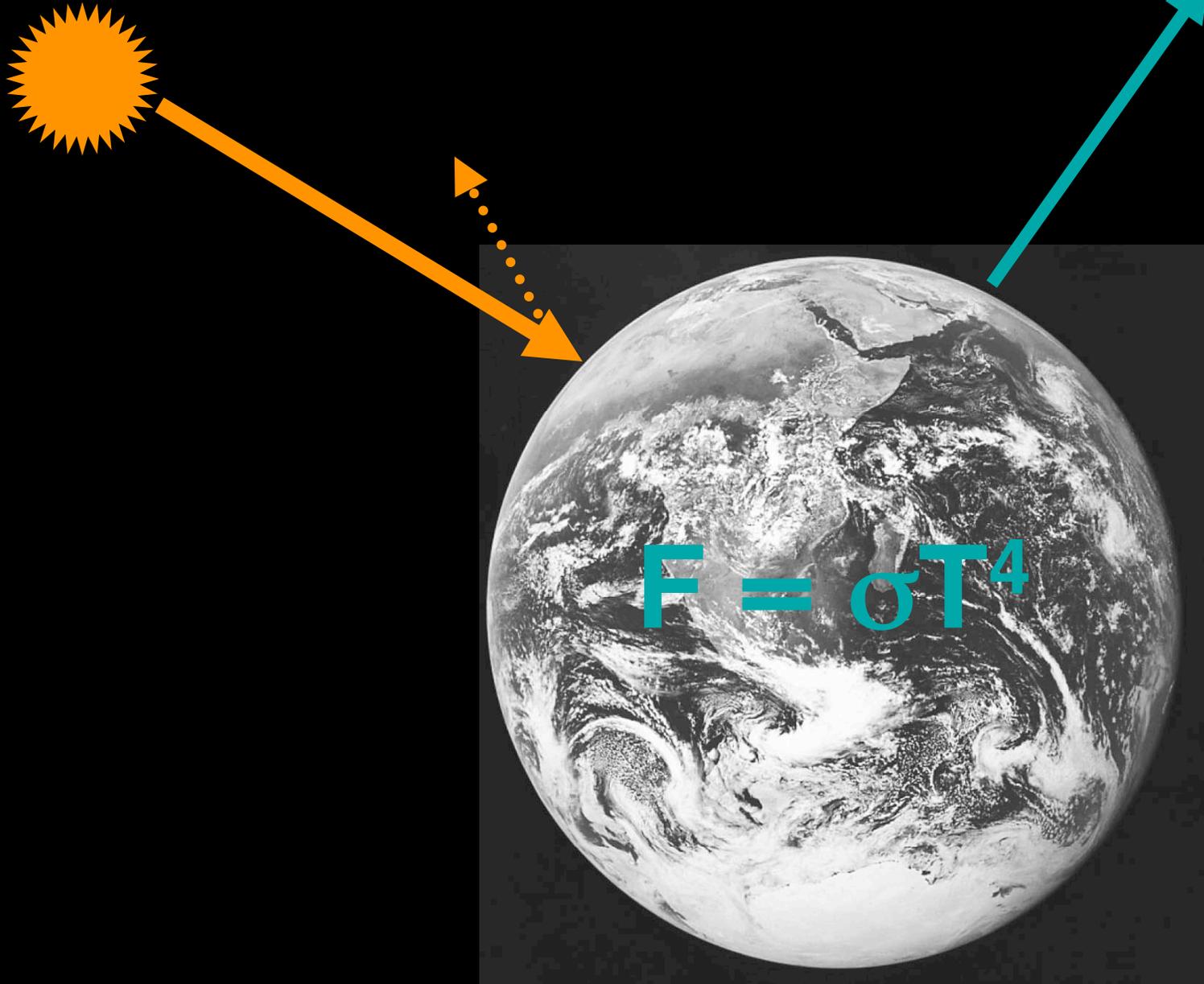


IV. Energy Balance



$$F = \sigma T^4$$

reminders from last class

- the Earth approximates a “blackbody”, both absorbing and emitting radiation
- the Earth receives incoming radiation from the Sun at relatively short wavelength (high energy)
- the Earth emits radiation at relatively long wavelengths (lower energy)
- Earth’s incoming and outgoing radiation are approximately in balance
- everything emits, i.e.:

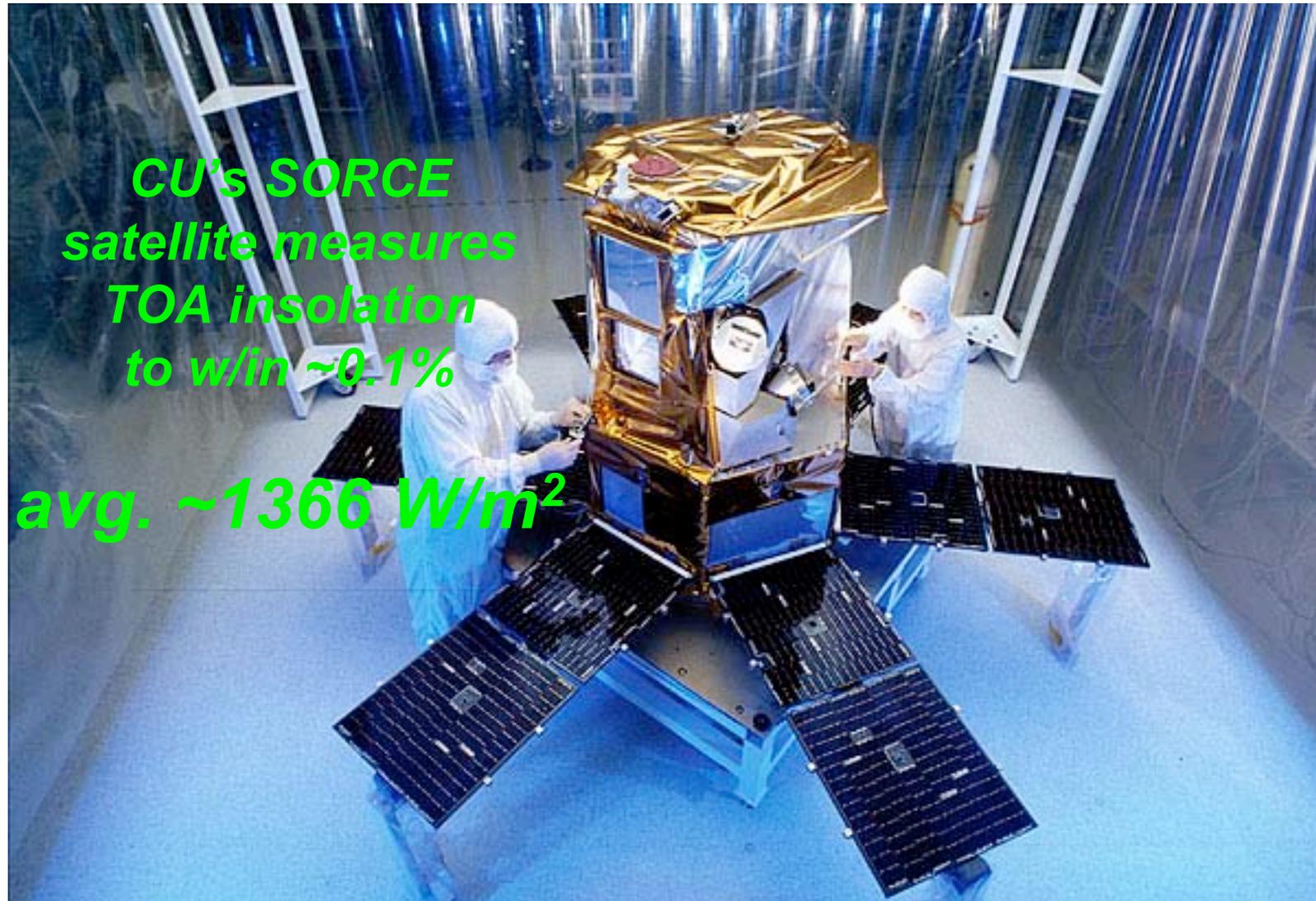
$$F = \sigma T^4$$

learning goals

- fates of incoming and outgoing radiation (transmission, reflection, and absorption)
- the concept of albedo (fractional reflectivity of a surface) incl. typical values of important surfaces
- the concept of energy balance
- mechanisms of selective absorption of radiation in the atmosphere
- knowledge of the main “Greenhouse gases”
- understanding of the origin and magnitude of the natural Greenhouse Effect

fates of incoming radiation

- *last class we described solar radiation reaching the top of Earth's atmosphere*



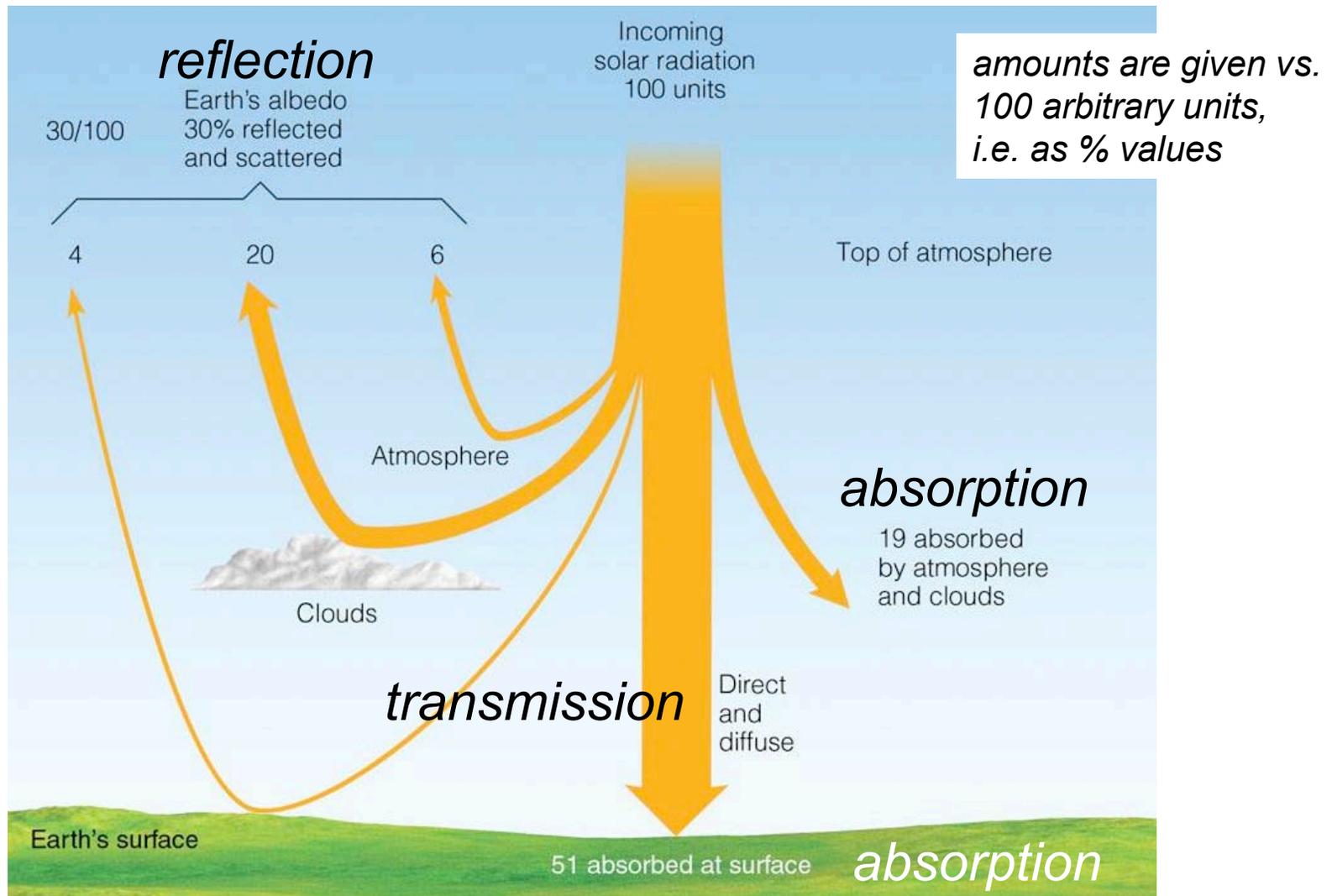
fates of incoming radiation

- *last class we described solar radiation reaching the top of Earth's atmosphere*

what happens next?

fates of incoming radiation

about half is transmitted thru the atmosphere to the surface, and.....



fates of incoming radiation

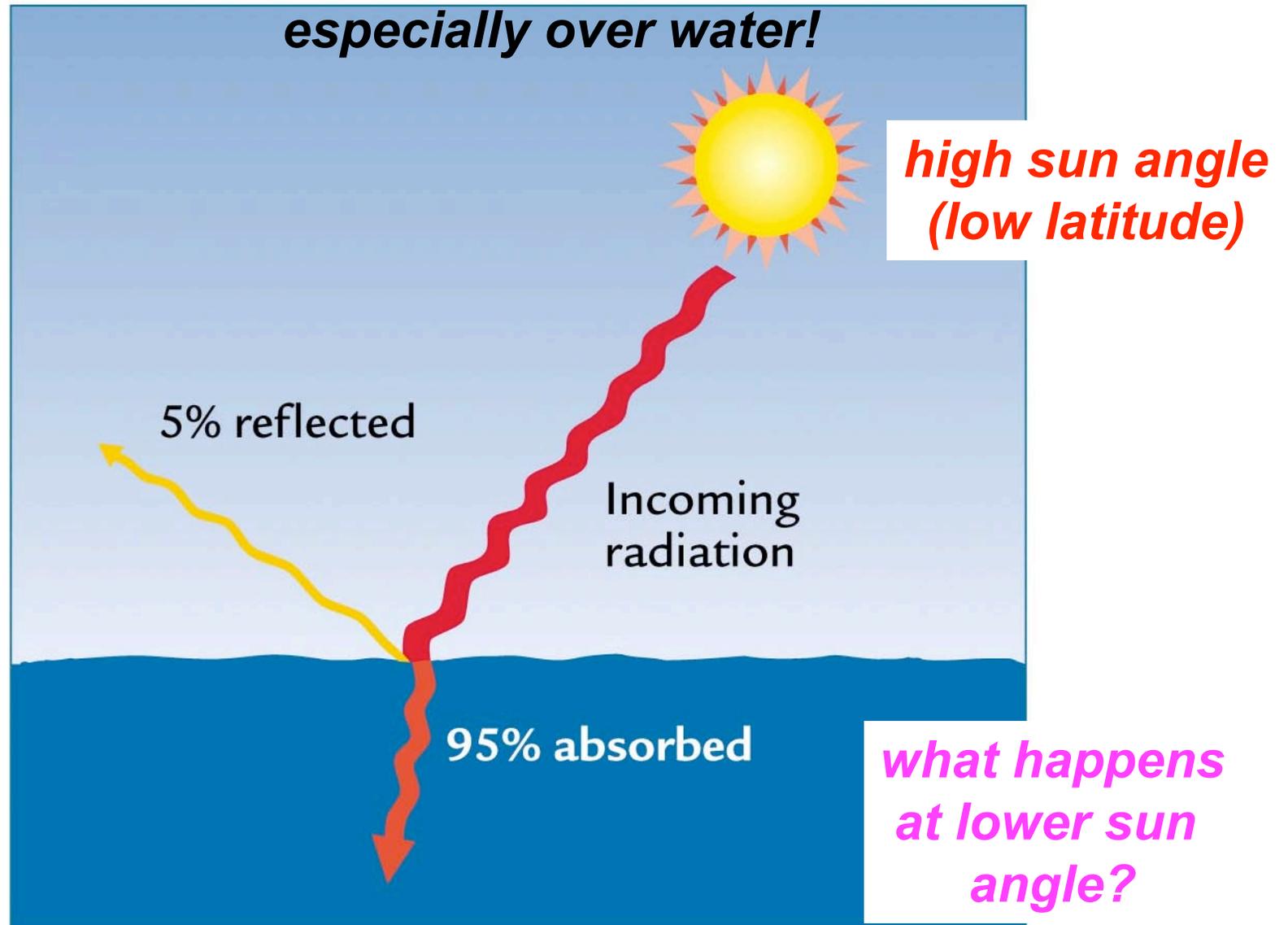
- ~30% of incoming radiation is *reflected* back to space
- this *fractional reflectivity* is called the *albedo*
- the *global* or *planetary albedo* is the fractional reflectivity of the Earth system as a whole
- in fact, all components of the Earth system have their own albedo
- what is the albedo of some typical components?

albedo (fractional reflectivity, α)

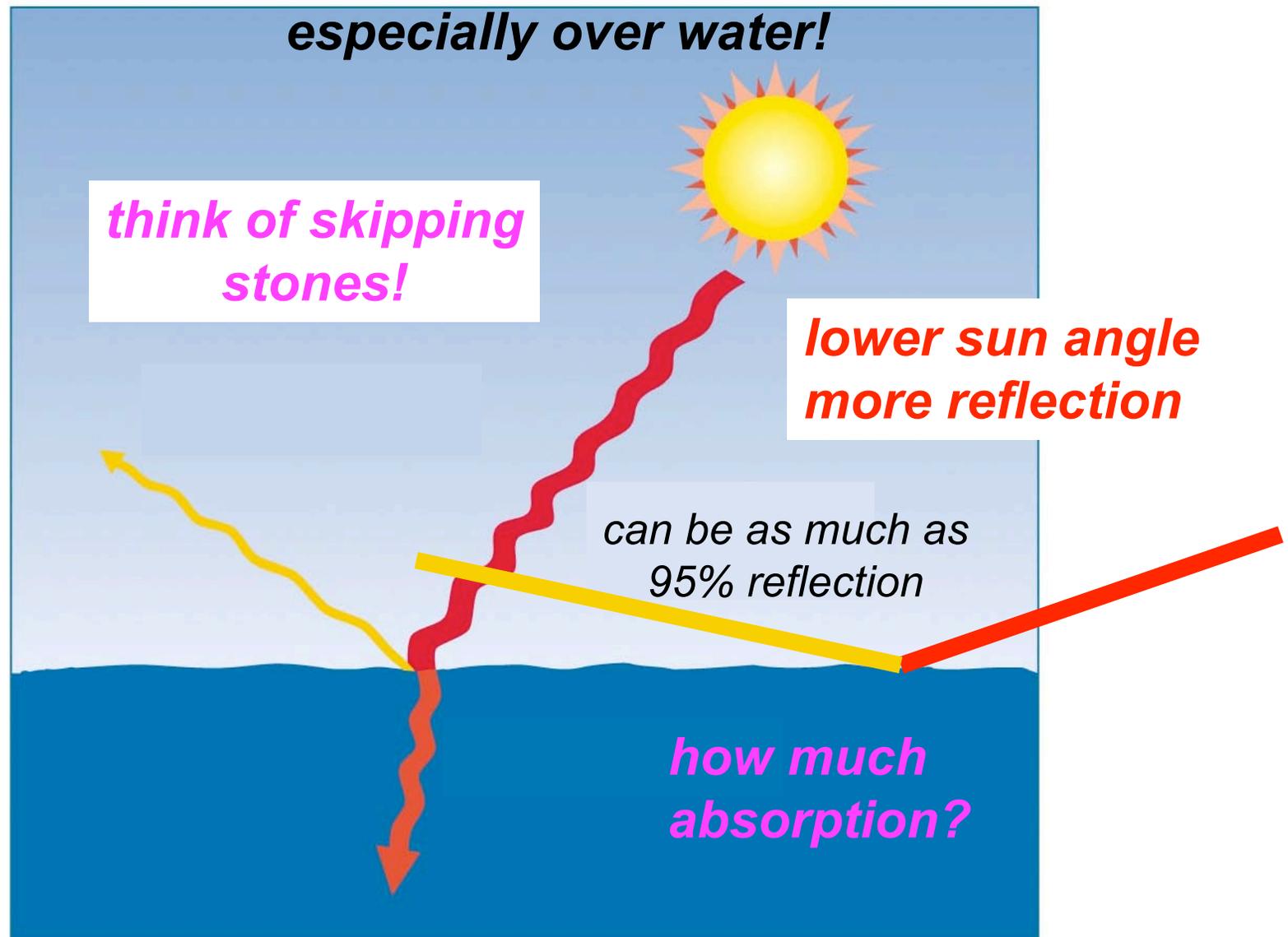
<i>Type of Surface</i>	<i>Albedo</i>
Sand	0.20–0.30
Grass	0.20–0.25
Forest	0.05–0.10
Water (overhead Sun)	0.03–0.05
Water (Sun near horizon)	0.50–0.80
Fresh snow	0.80–0.85
Thick cloud	0.70–0.80
Low thin cloud	0.30–0.50

lighter surfaces have higher albedo than darker ones

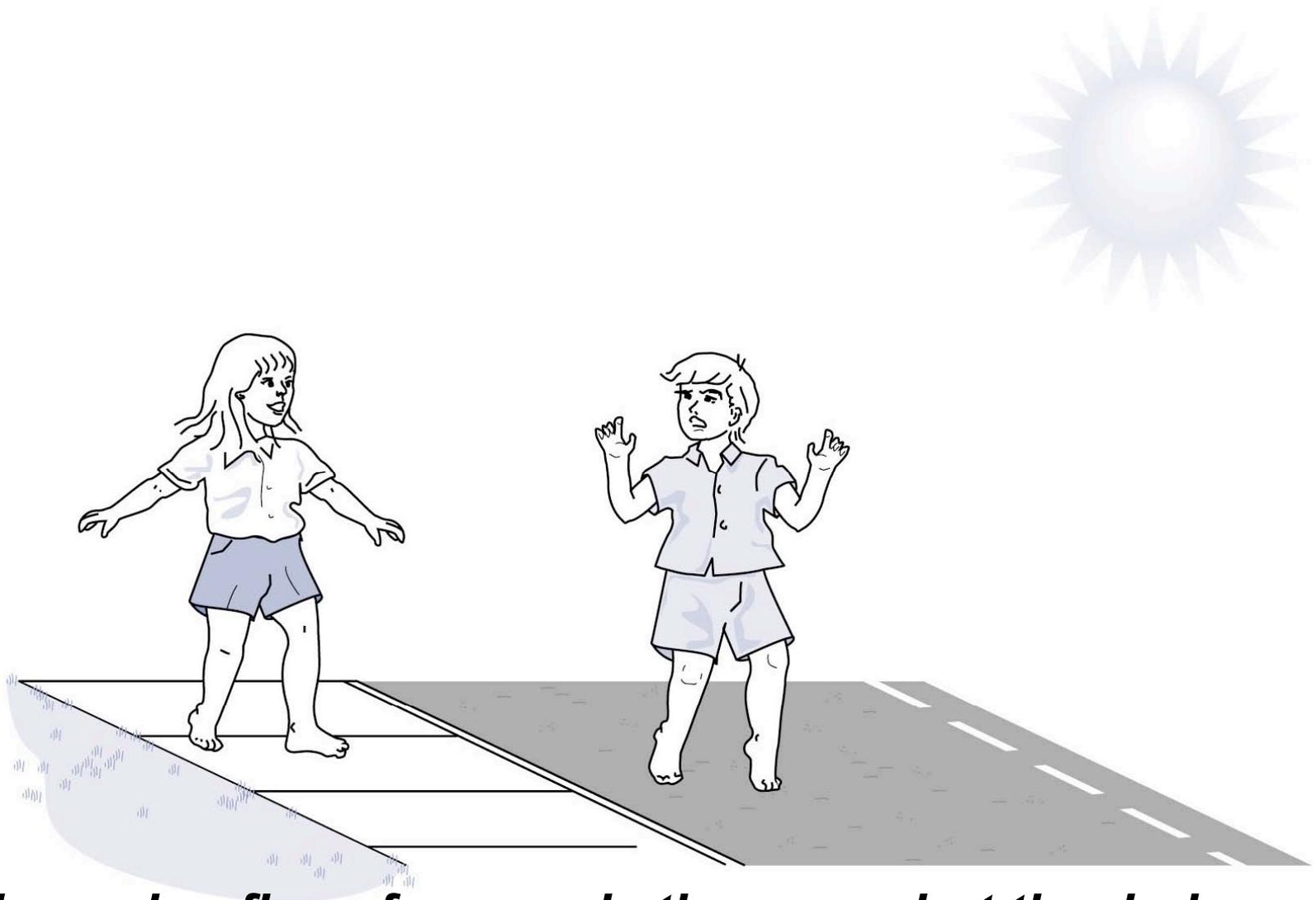
sun angle influences albedo



sun angle influences albedo



does albedo matter?

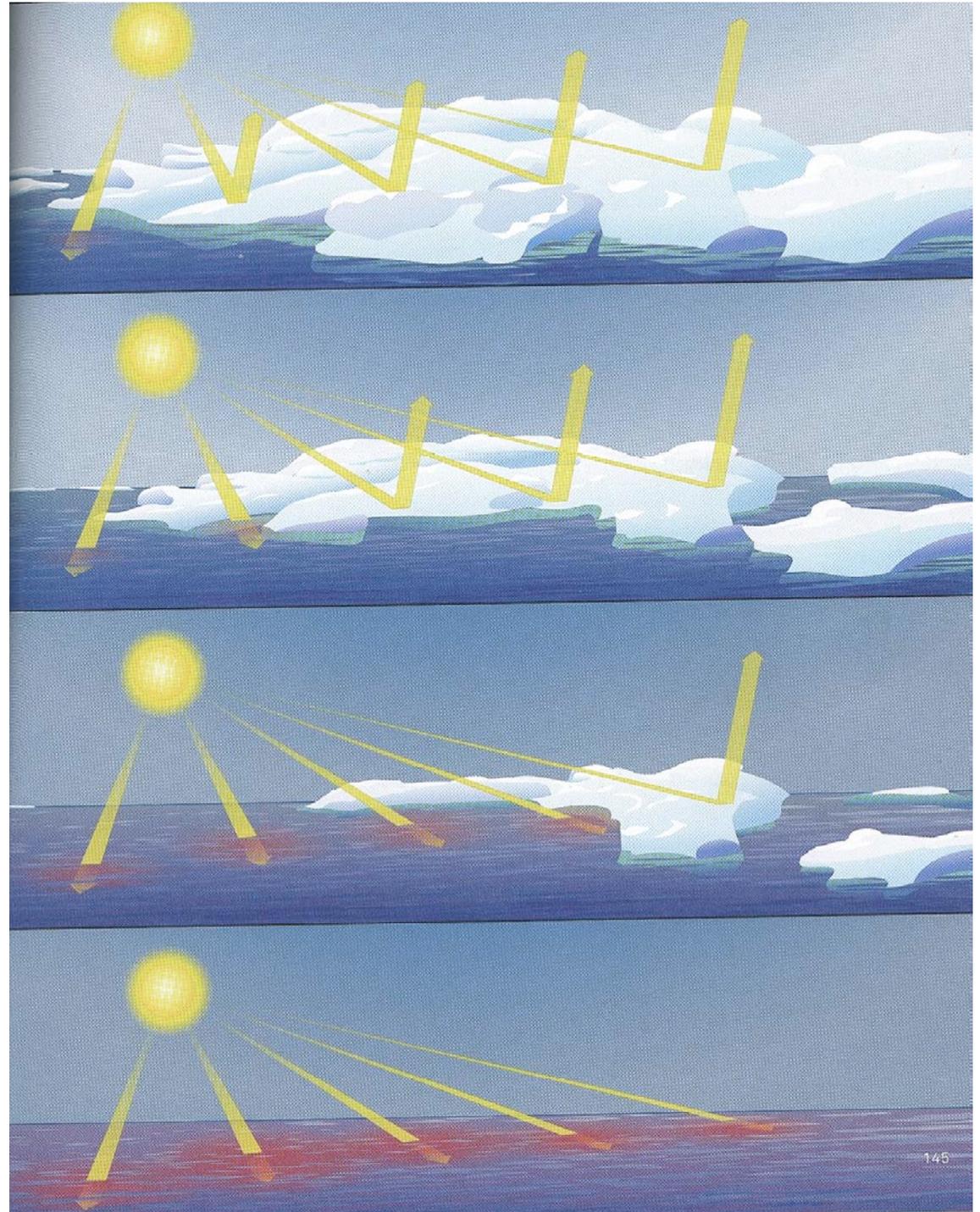


the incoming flux of energy is the same, but the darker surface is hotter

*recall feedback
process discussed
last week ?*

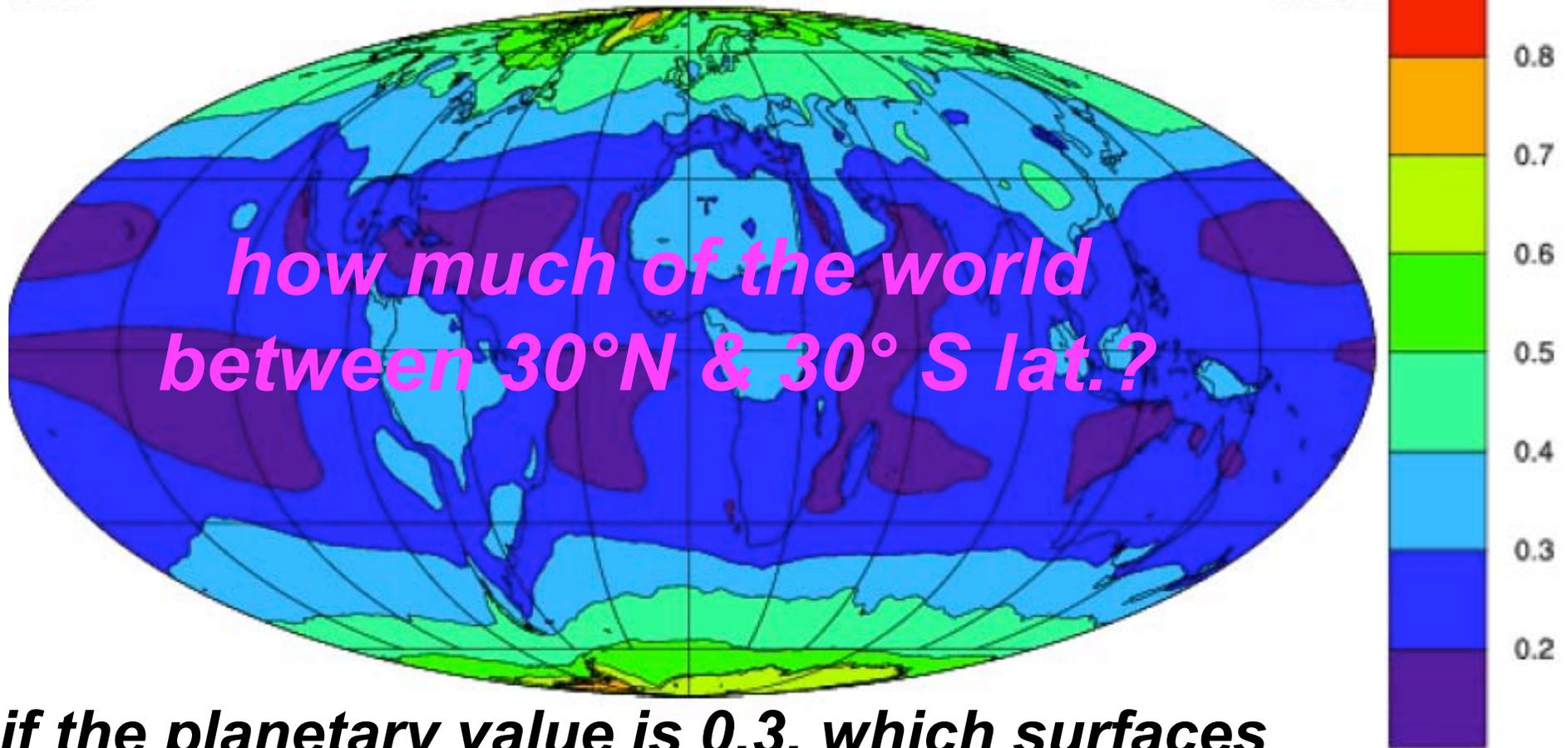
*sea ice is bright
and reflects ~90%
of the suns rays,
sea water is dark
and absorbs ~90%
of the suns rays.*

source: A. Gore's "AIT"



global annual mean albedo fraction

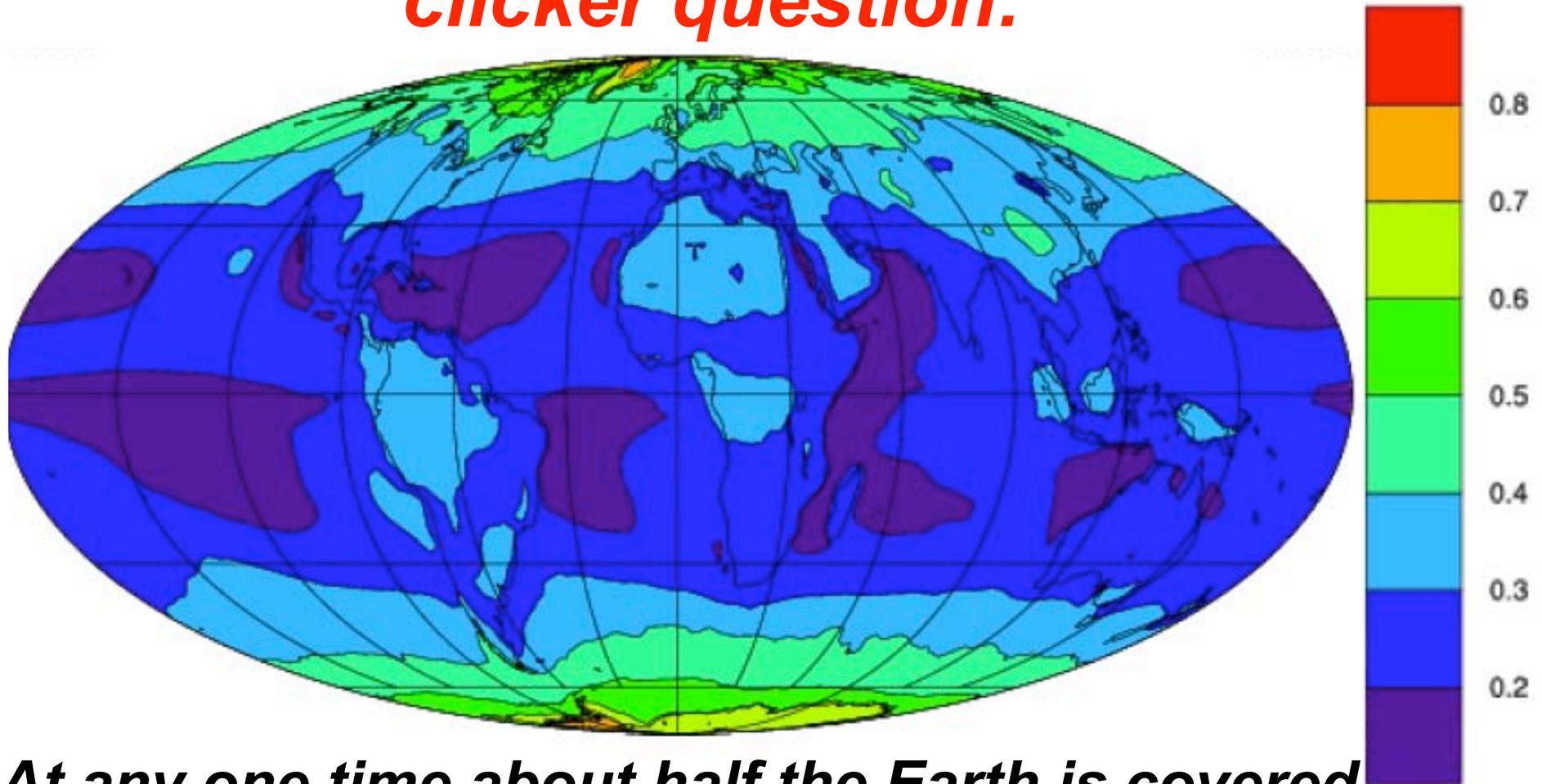
clicker question:



if the planetary value is 0.3, which surfaces contribute most by area to this average value a) ice, b) snow, c) vegetated ground, d) oceans & clouds, e) deserts

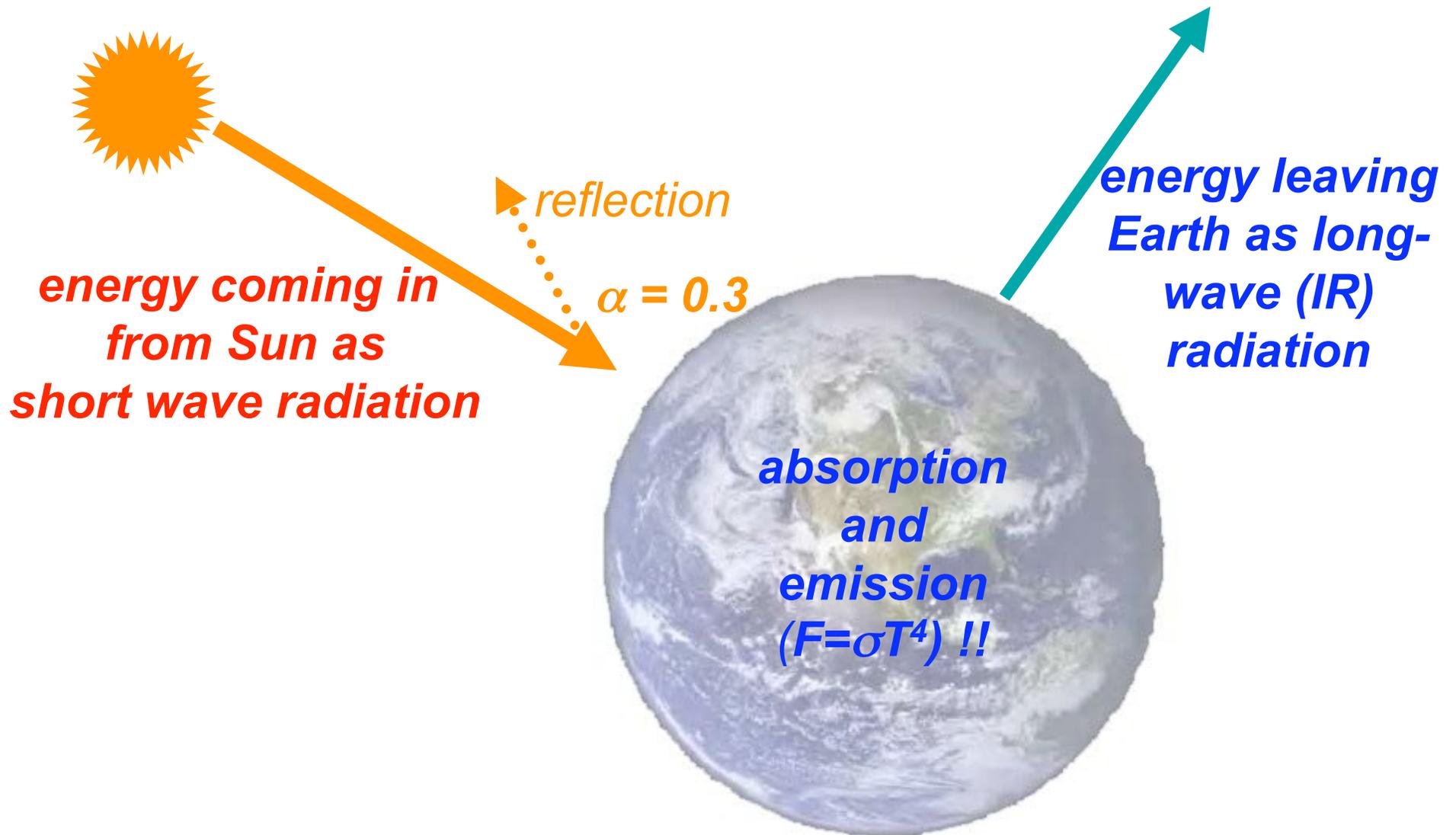
global annual mean albedo fraction

clicker question:



At any one time about half the Earth is covered in cloud. For this reason the annual average albedo of the two hemispheres is about the same despite the large difference in land area.

energy balance



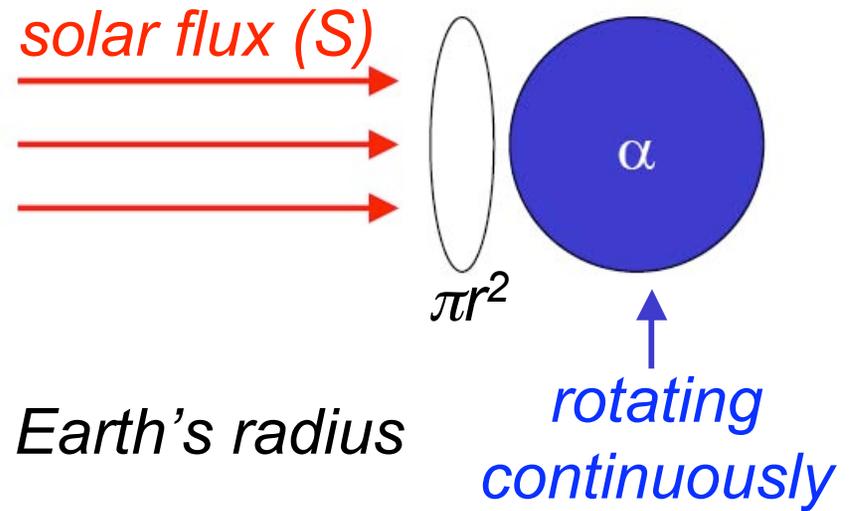
Earth's temperature will adjust so that fluxes of emitted and absorbed radiation will balance

radiative balance

the earth viewed from sun is a disc, so the solar input of energy is:

$$IN = S(1-\alpha) \times \pi r^2$$

r is Earth's radius



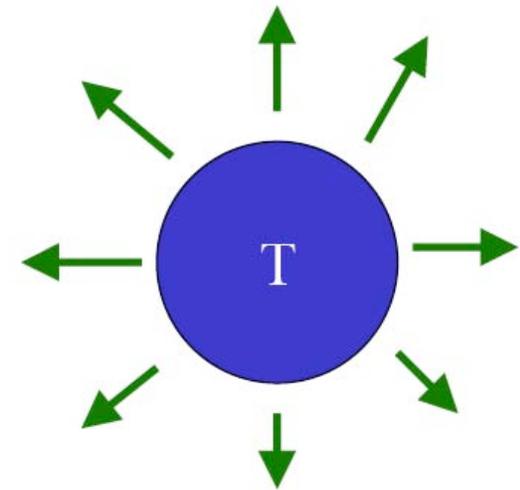
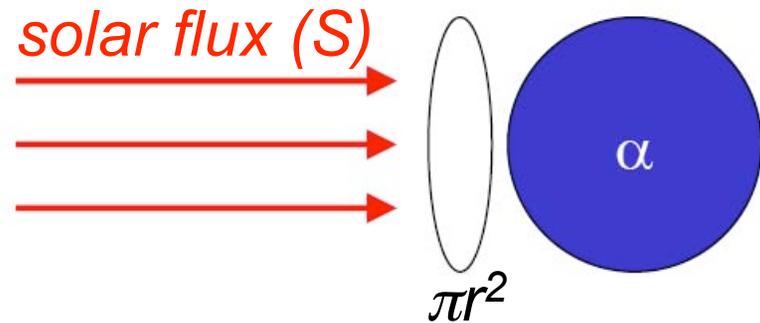
radiative balance

the earth viewed from sun is a disc, so the solar input of energy is:

$$IN = S(1-\alpha) \times \pi r^2$$

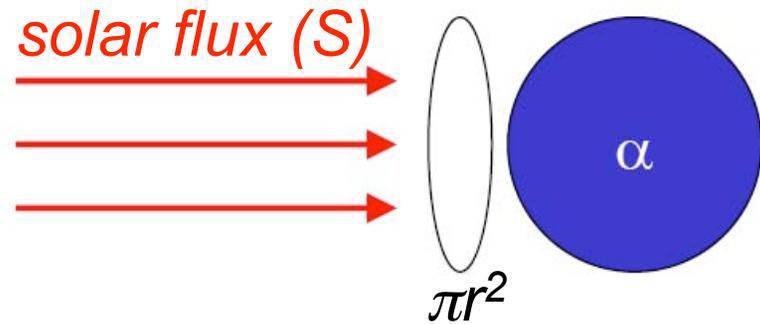
but the Earth as a blackbody radiates in all directions according to its temperature, so the energy out is:

$$F = \sigma T^4 \times 4\pi r^2$$
$$OUT = \sigma T^4 \times 4\pi r^2$$



Earth radiating in all directions, i.e. $4\pi r^2$

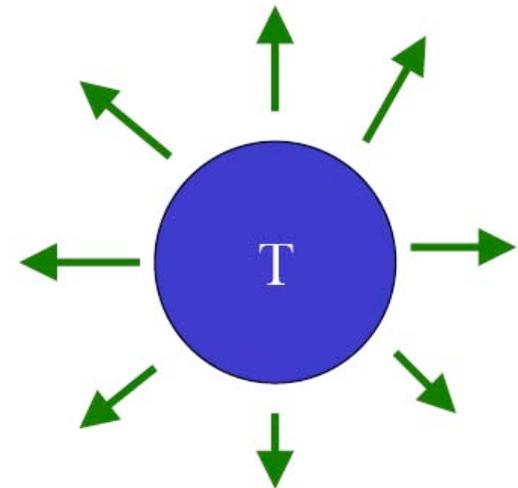
radiative balance



$$IN = OUT$$

$$S(1-\alpha) \times \cancel{\pi r^2} = \sigma T^4 \times \cancel{4\pi r^2}$$

$$S(1-\alpha)/4 = \sigma T^4$$



Earth radiating
in all directions,
i.e. $4\pi r^2$

effective radiation temperature

- *we now have everything we need to determine what the Earth's temperature should be given the amount of radiation coming in from the Sun*
- *S , the solar radiation flux at the top of the atmosphere is 1360 W/m^2*
- *α , the planetary albedo is 0.3*
- *σ , the Stefan Boltzmann constant is known (and is in your text!)*
- *using $S(1-\alpha)/4 = \sigma T^4$, we can re-arrange the equation and solve for T , the temperature (on the Kelvin scale)*

effective radiation temperature

the calculated result is

255 K or -18 °C

***what do you think the actual
measured temperature of the
Earth surface is??***

effective radiation temperature

the calculated result is

255 K or -18 °C

but the measured temperature

is 288 K or +15 °C

33 °C (K) higher?

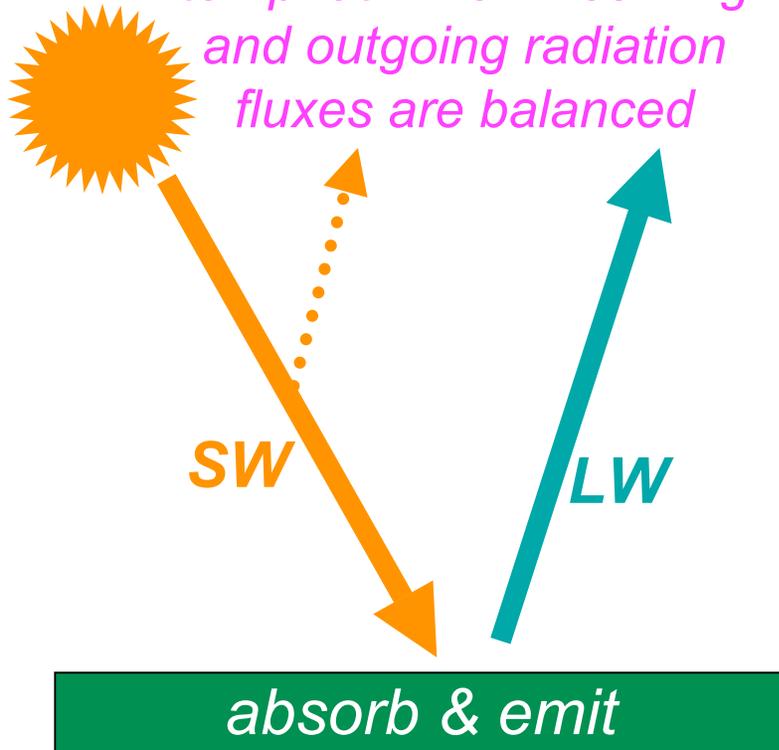
How can this be?

welcome to the greenhouse!

- this mathematical exercise demonstrates the existence of the *natural “Greenhouse Effect”*
- as we shall now see, it is due to the envelope of atmospheric gases that absorb and re-emit radiation at specific wavelengths
- without it, the earth would be too cold to sustain life as we know it.....

welcome to the greenhouse!

*Stefan Boltzmann
Law helps us describe
temp. at which incoming
and outgoing radiation
fluxes are balanced*

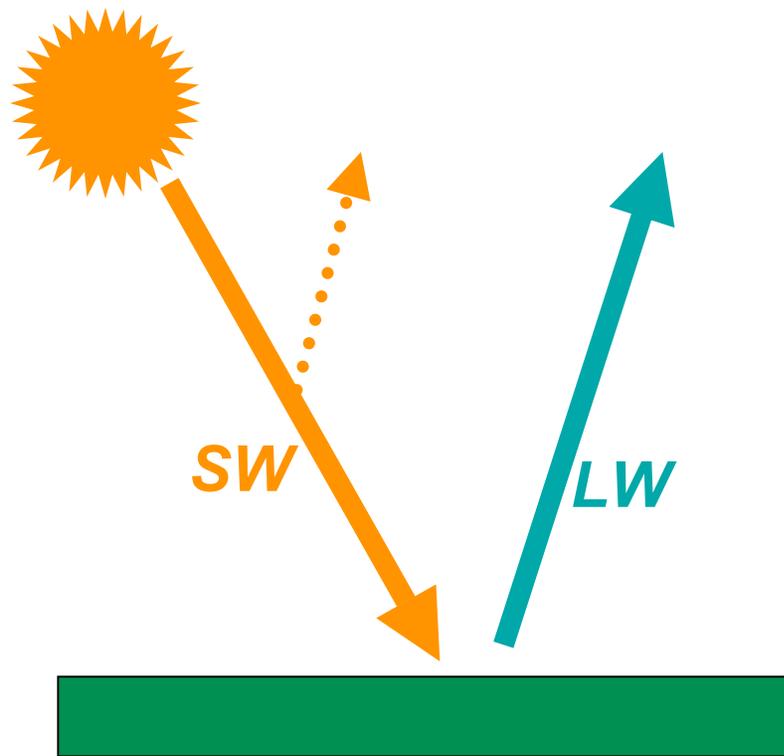


$$F = \sigma T^4 !!$$

w/o atmosphere

$$T = -18 \text{ }^\circ\text{C}$$

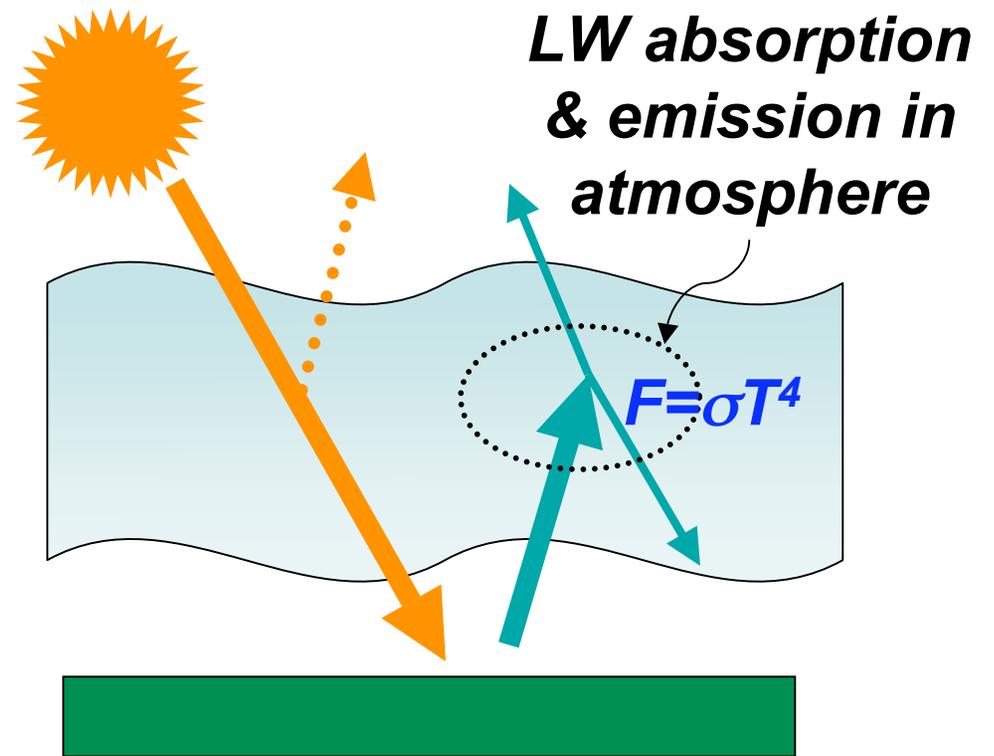
welcome to the greenhouse!



$$F = \sigma T^4 !!$$

w/o atmosphere

$T = -18 \text{ }^\circ\text{C}$

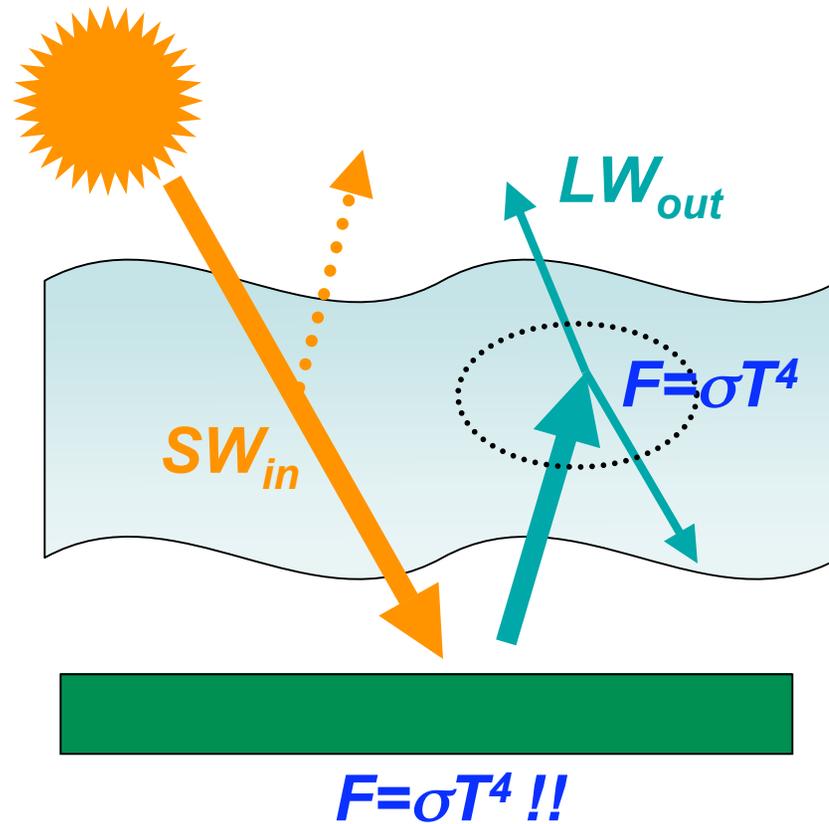


$$F = \sigma T^4 !!$$

w/ atmosphere

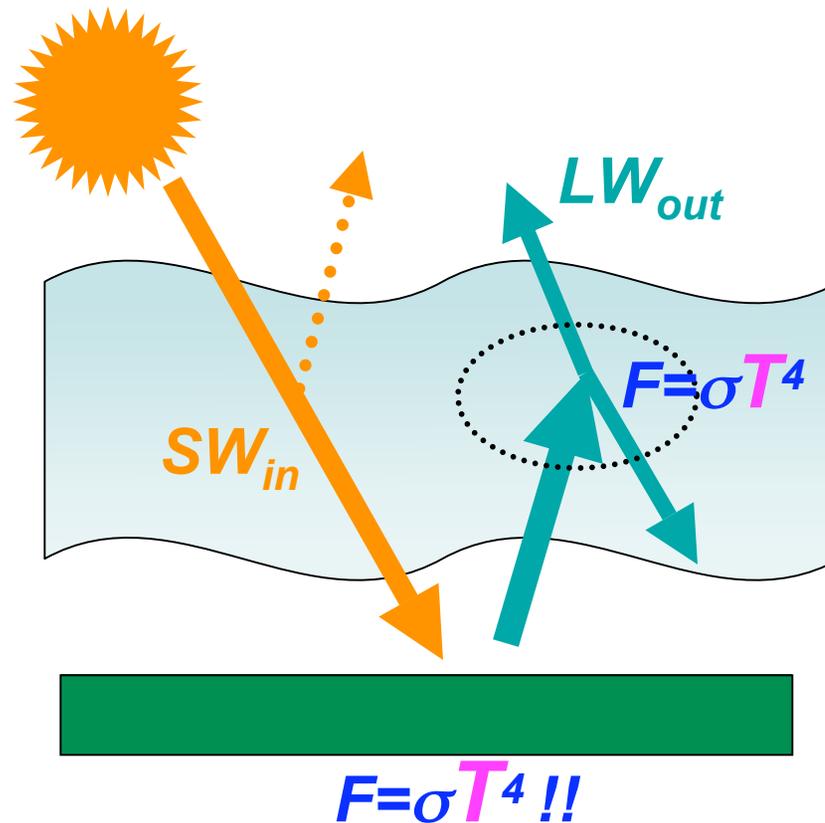
$T = 15 \text{ }^\circ\text{C}$

what must happen?



***the incoming flux of SW radiation and outgoing flux of LW radiation are briefly out of balance
what must happen to restore balance?***

what must happen?



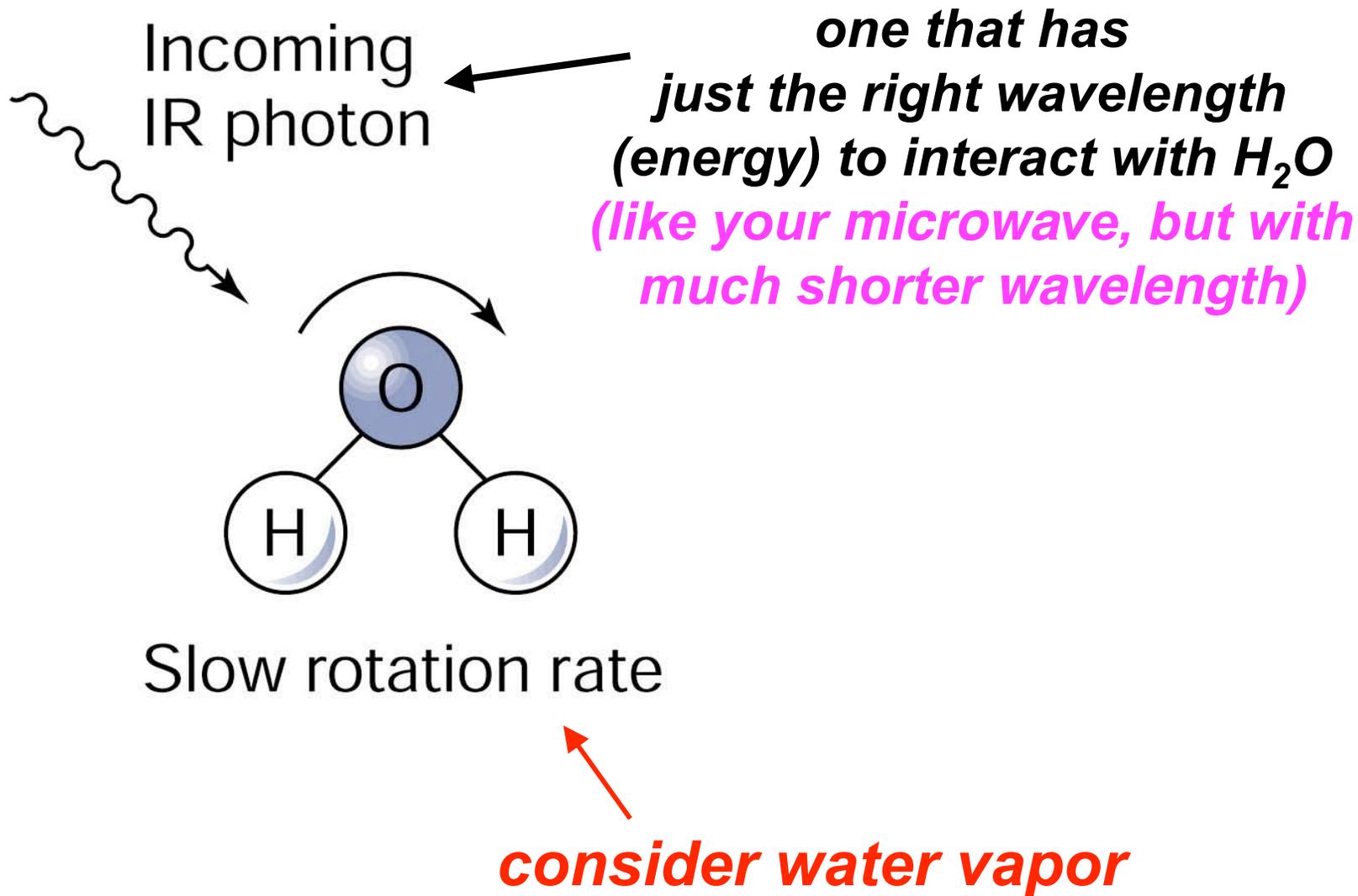
toggle
this!

***the incoming flux of SW radiation and outgoing
flux of LW radiation are briefly out of balance
 T must go up = Greenhouse Warming***

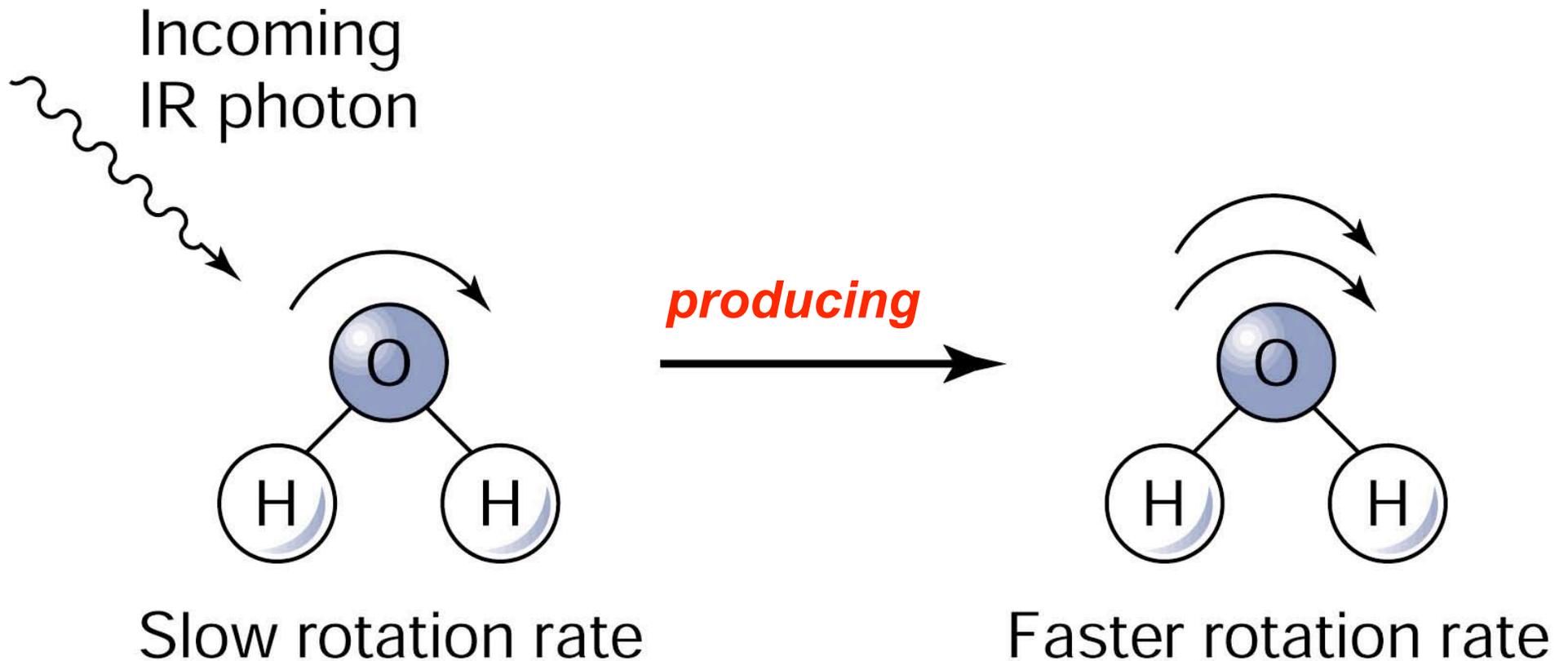
what is a greenhouse gas?

- unlike solids (blackbodies), gases are *selective* absorbers and emitters of radiation
- greenhouse gases are generally transparent to SW (incoming) radiation, but opaque to some LW Earth (outgoing) radiation

how does absorption work?

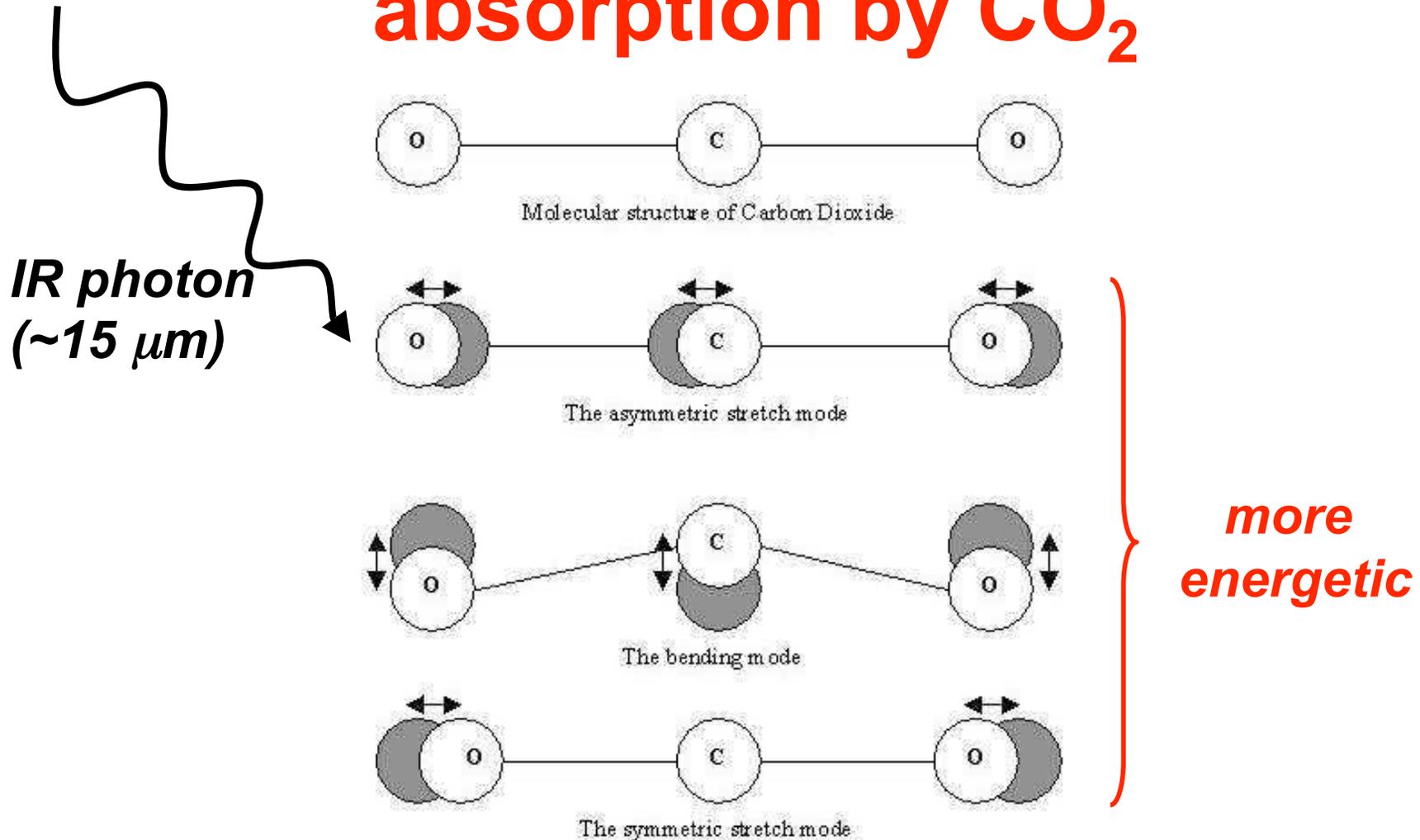


how does absorption work?



*the now faster molecule has more energy,
and can re-emit radiation to loose that energy*

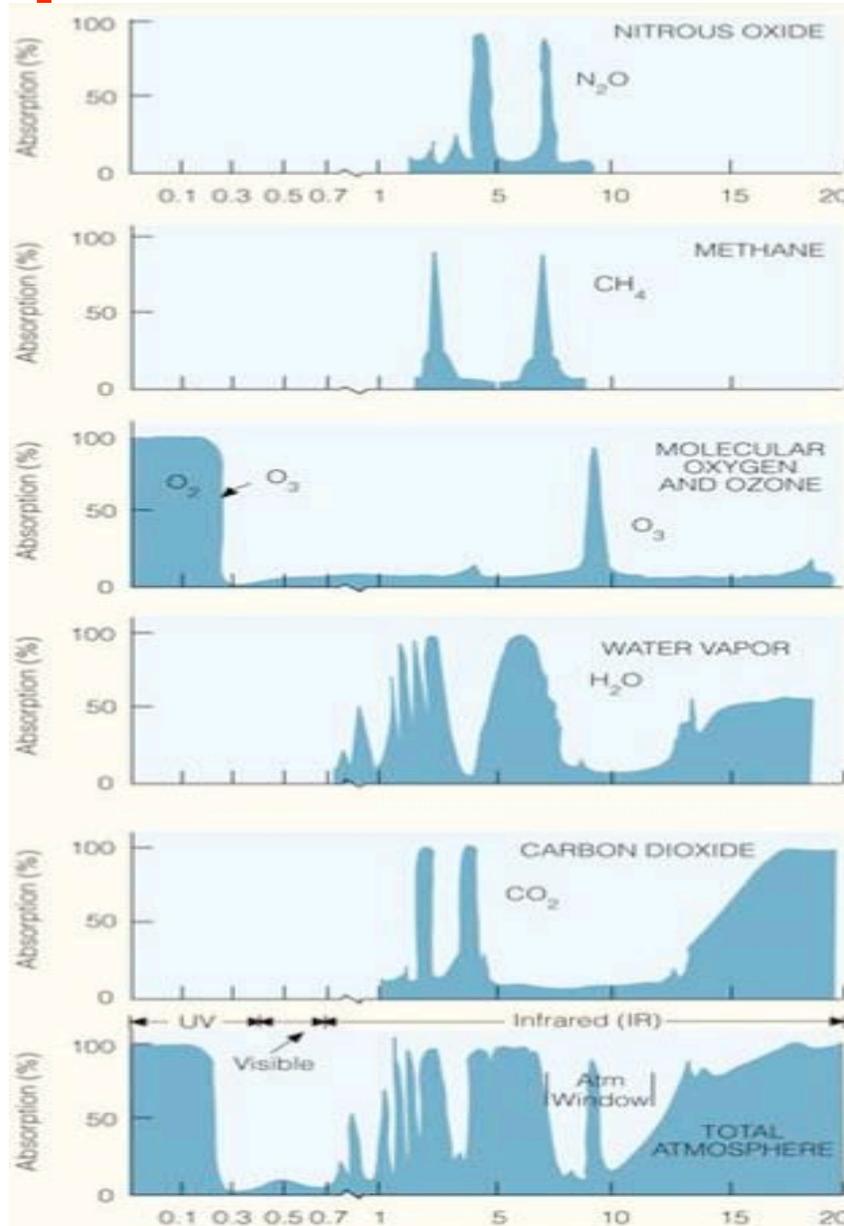
absorption by CO₂



*photon of select “wavelength” is absorbed to produce **vibration** and **bending**, adding energy that can be **re-emitted** as LW radiation*

absorption in the atmosphere

% radiation absorbed



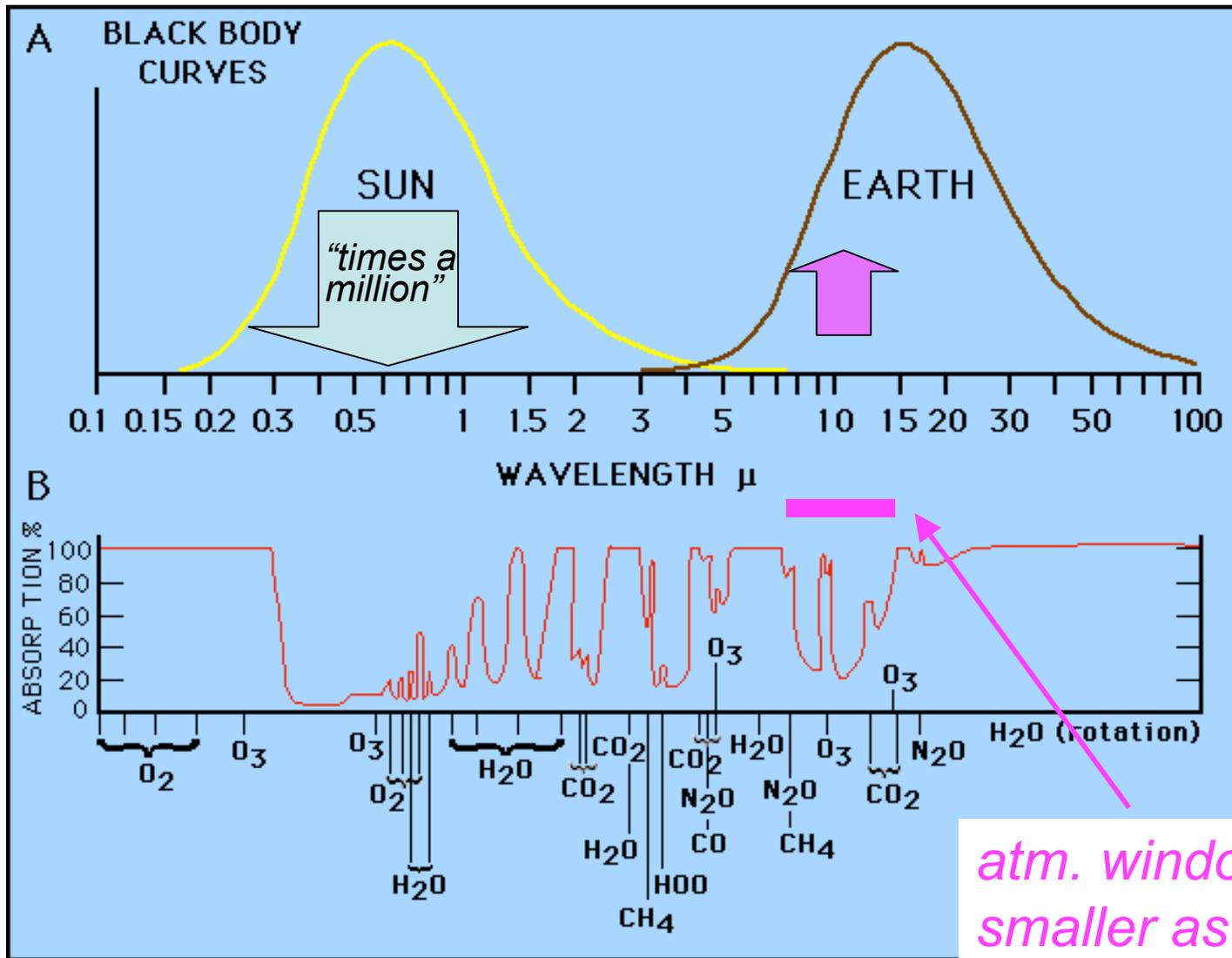
what is this graph telling us?

shorter

wavelength (μm)

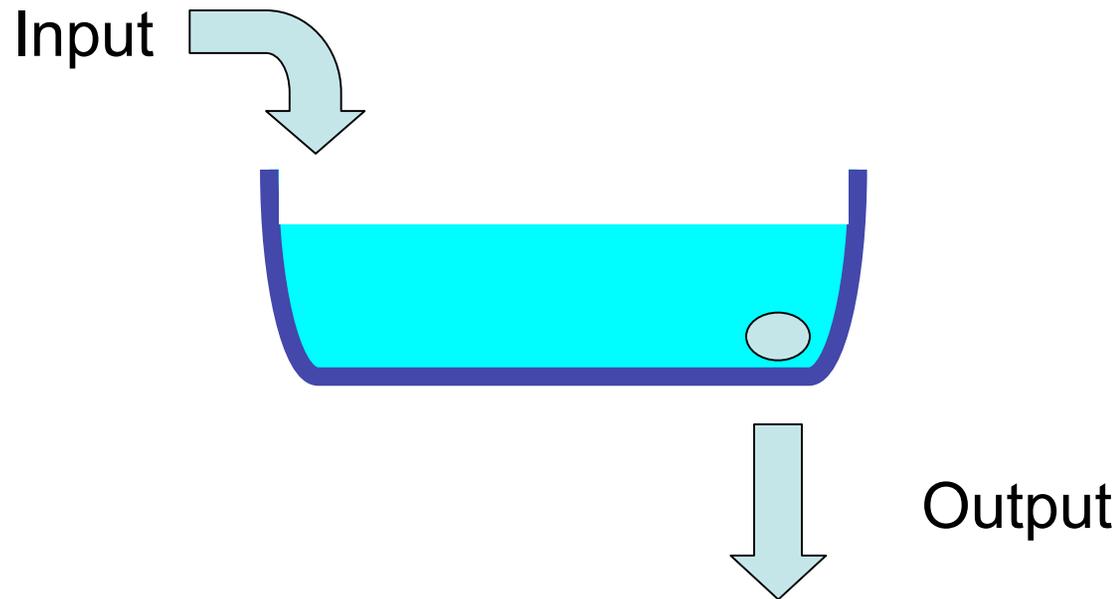
longer

absorption in the atmosphere



atm. window gets smaller as GHGs are added

bath tub

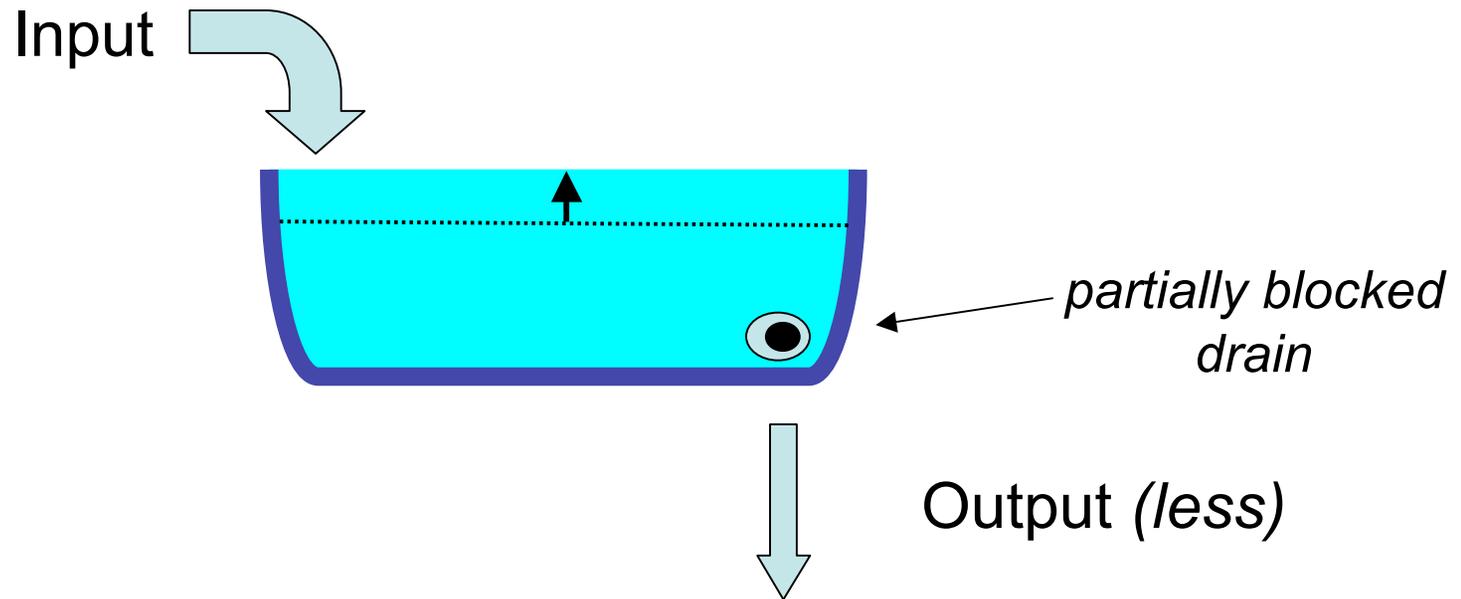


when the flow of water into the tub equals the flow out of the tub, the water level does not change

steady state conditions:

$$\text{input} = \text{output}$$

“greenhouse” bath tub

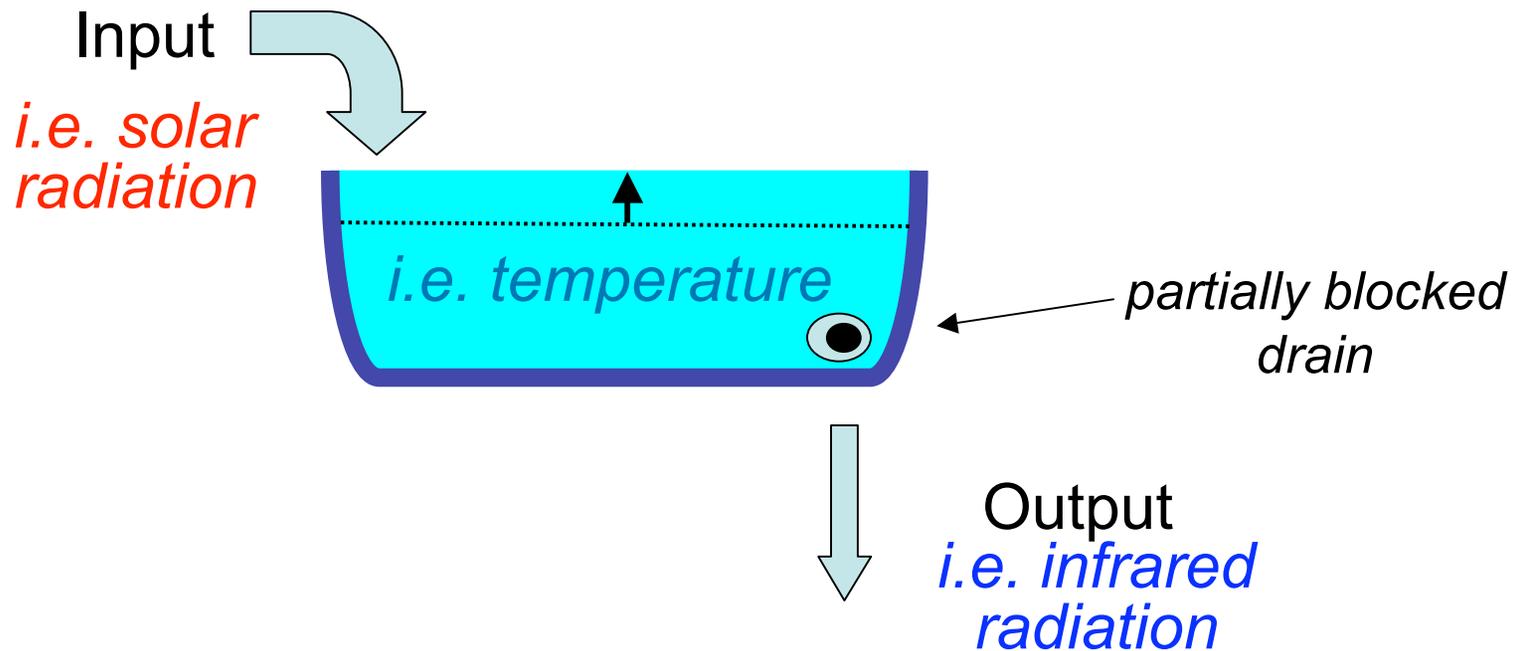


***when the flow of water into the tub
exceeds the flow out of the tub, the
water level rises***

***in this system the water level rises
forever***

what happens in the climate system?

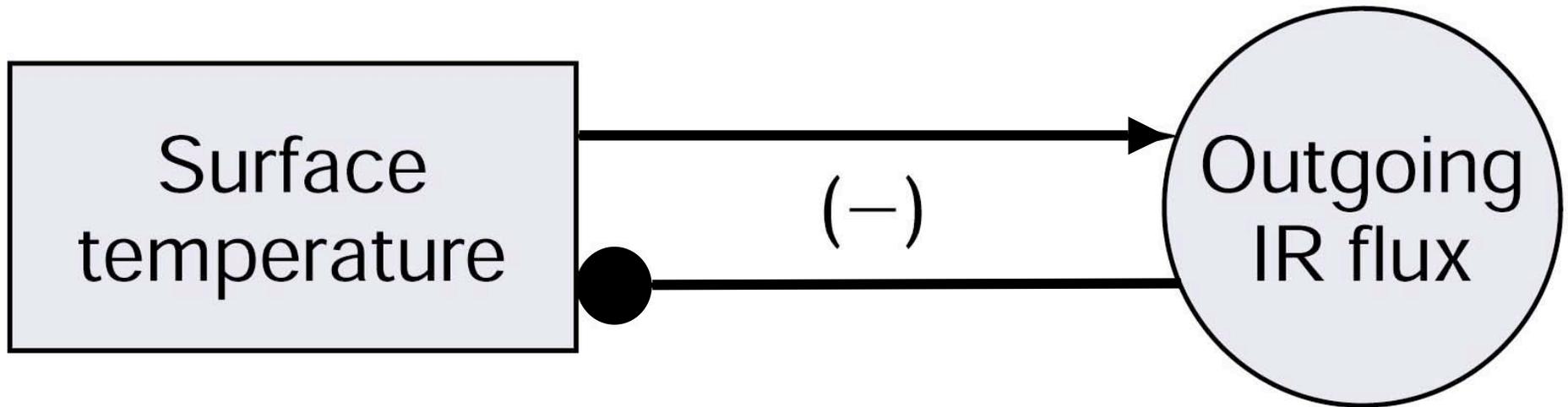
clicker question



The water in this system rises forever. Using the analogy to the climate system we have developed, what process would allow the water depth to stabilize at a new level?

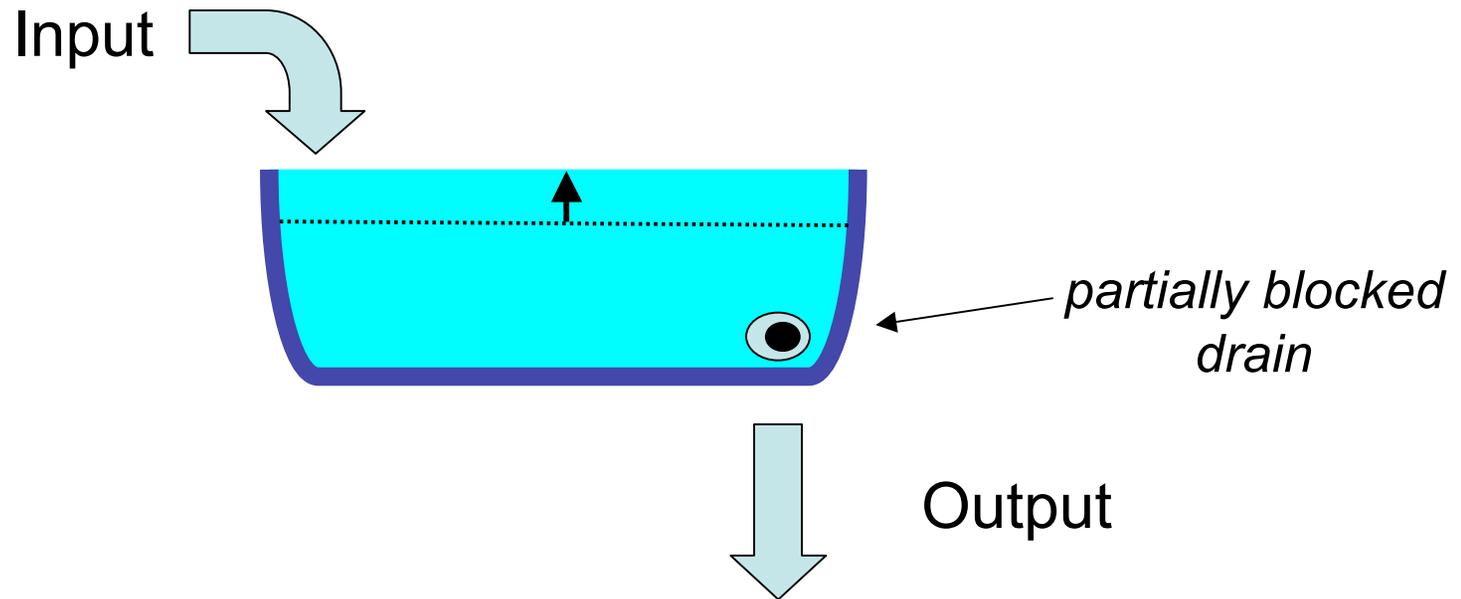
a) the input would adjust downward, b) the output would adjust upward, c) the temperature-infrared feedback would kick in, d) both a and c, e) both b and c.

mother of all feedbacks



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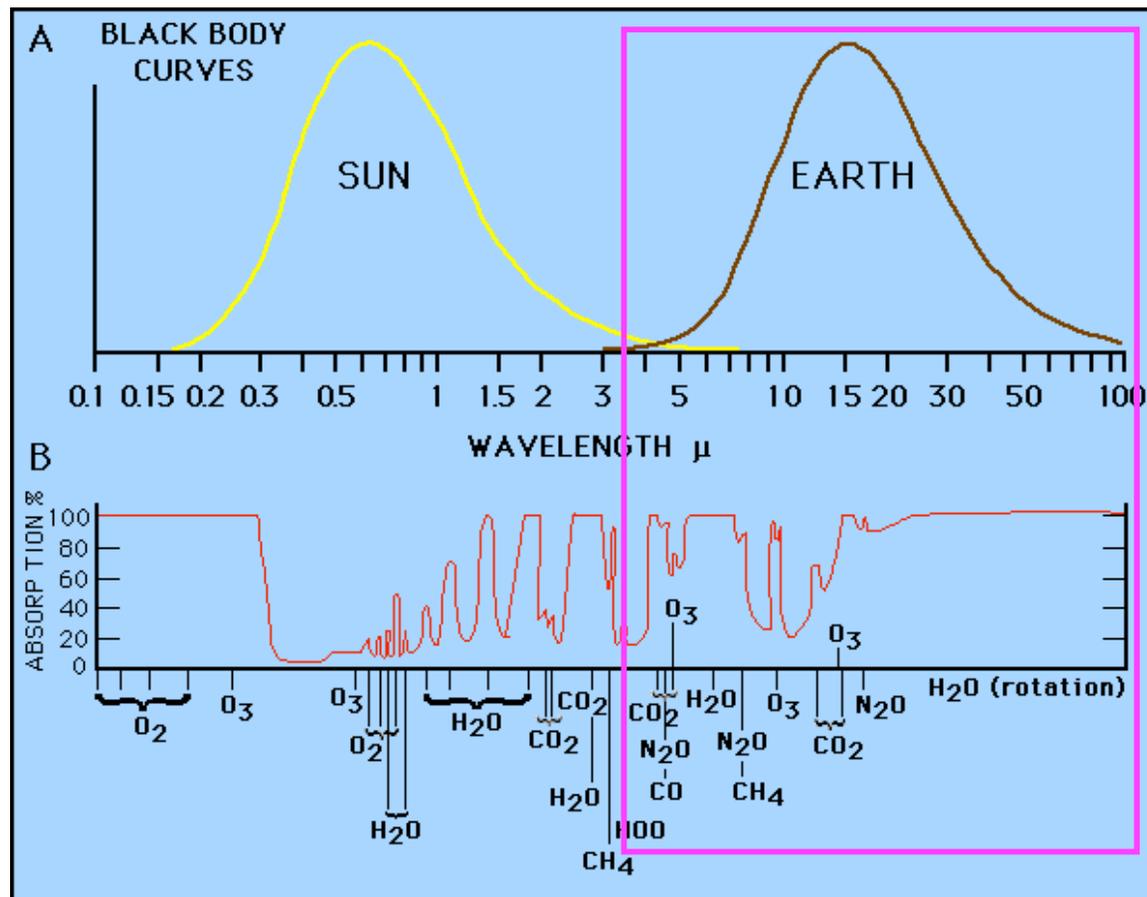
“greenhouse” bath tub



***the output “pressure” effectively goes up through the operation of the temperature - infrared feedback
($F = \sigma T^4!$)
but the new water level
(i.e. temperature) is now higher than
before***

the real deal....

let's now use an on-line radiation model to see how the greenhouse really works....



Go to: <http://forecast.uchicago.edu/Projects/modtran.html>

In class we used the textbook author's online radiation model in order to demonstrate

- 1) how the addition of greenhouse gases influences the spectrum and amount of outgoing infrared radiation and
- 2) how changing surface temperature influences the amount of outgoing infrared radiation.

We found that we needed to raise surface temperature in order to maintain balance between incoming and outgoing radiation after greenhouse gases were added to the model atmosphere.

clicker question:

Which of the following gases would
ABSORB radiation

- A. Water vapour (H_2O)
- B. Carbon dioxide (CO_2)
- C. Ozone (O_3)
- D. All of the above
- E. None of the above

clicker question:

Which of the following gases would
EMIT radiation

- A. Water vapour (H_2O)
- B. Carbon dioxide (CO_2)
- C. Ozone (O_3)
- D. All of the above
- E. None of the above

key points

- we have demonstrated the magnitude and mechanism of the natural greenhouse effect!
- we have described the major heat flows within the global energy budget!
- ***next week:***
- we begin with a look at ***atmospheric composition and structure***
- reading: Ch 4 pp. 44-53 & Ch 19 pp 381-2.
- ***Homework: posted tonight, due Tuesday!***